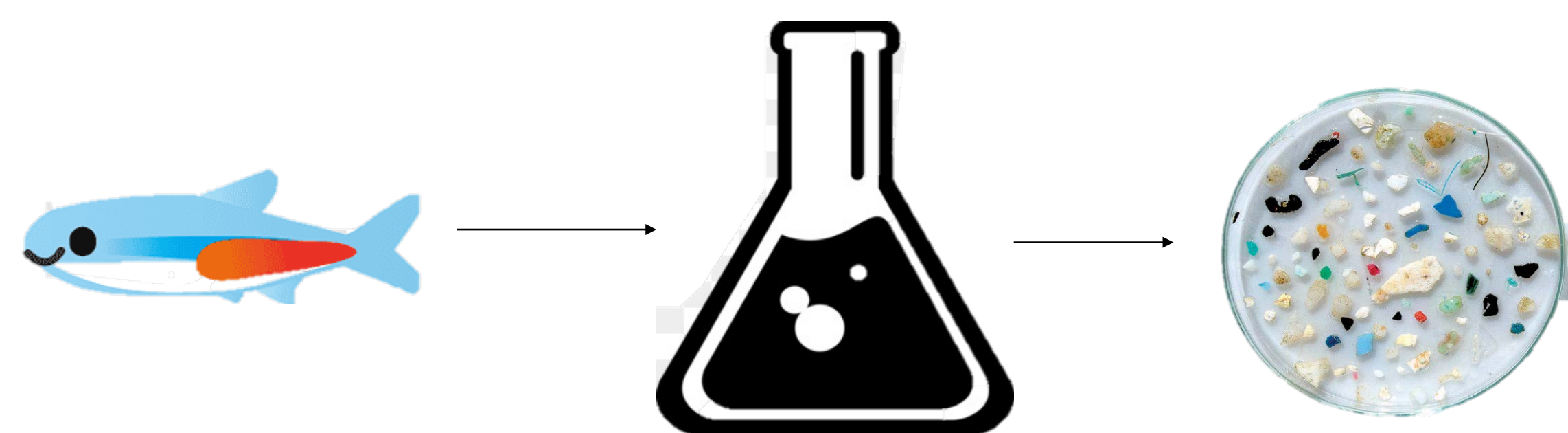


1. Introduction

Chemical digestion methods allow plastics to be extracted and quantified from animal biota during field collection and exposure studies. However, to efficiently remove biomass, these methods often utilise harsh chemical conditions which can damage the plastic polymer.



[pngwing.com](https://www.pngwing.com/); [favpng.com](https://www.favpng.com/); [sci360.org](https://www.sci360.org/)

To ensure accurate quantification and classification of plastics in the environment, suitable chemical digestion methods need to be implemented.

2. Methods

Prepared polystyrene microplastics (300 μm – 1 mm) were exposed to the following digestion conditions in the table below

Reagent	Time (hours)	Temperature (°C)
KOH	12, 24, 48	30, 60, 90
NaOH		
H ₂ O ₂		
HNO ₃	1, 2, 12	30, 60

3. Results and Discussion

☐ Treatments with HNO₃ resulted in a significant decrease in polymer chain length (e.g. molecular weight), changes to signature peak environments in the FT-IR spectra and severe mechanical damages indicating polymer degradation.

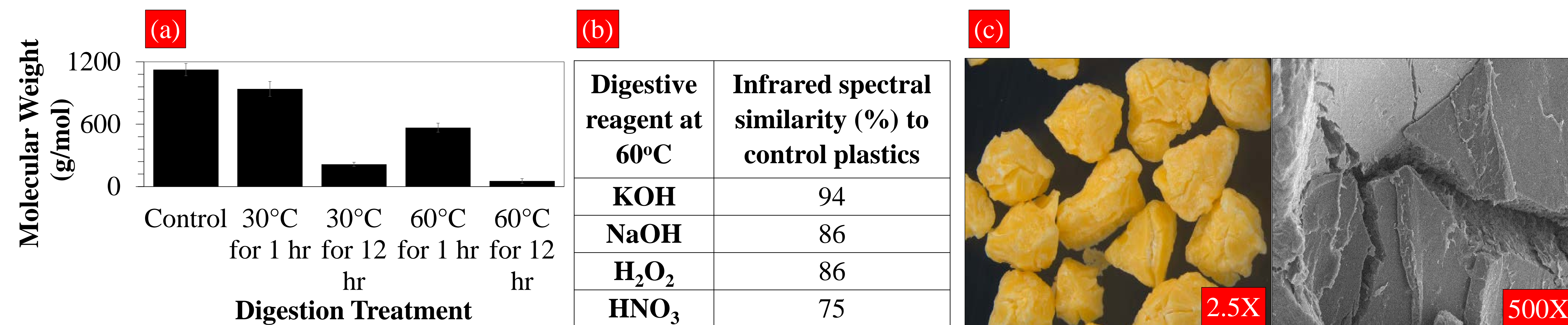


Figure 1. Chemical and physical analyses revealed (a) polymer degradation, (b) infrared spectral dissimilarity and (c) fragment discolouration and deep surface cracking associated with high temperature HNO₃ treatments

☐ There was significant swelling and pitting after all 90°C KOH, NaOH and H₂O₂ digestions. This indicated that microplastics may be more sensitive to digestion conditions given their manufacturing origins and/or environmental histories.



Figure 2. Microplastics swollen and pitted after 48-hour, 90°C digestions with (a) KOH, (b) NaOH and (c) H₂O₂ at 2.5X magnification

4. Conclusions

- ☐ Surface morphology may impact fragment reactivity during digestion.
- ☐ High temperatures digestions are associated with enhanced reagent reactivity, polymer and fragment degradation.
- ☐ HNO₃ digestions above 60°C may not be suitable for microplastic extractions